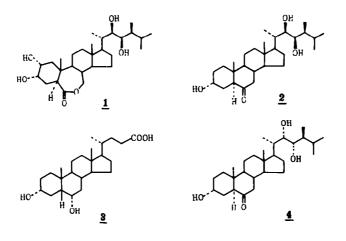
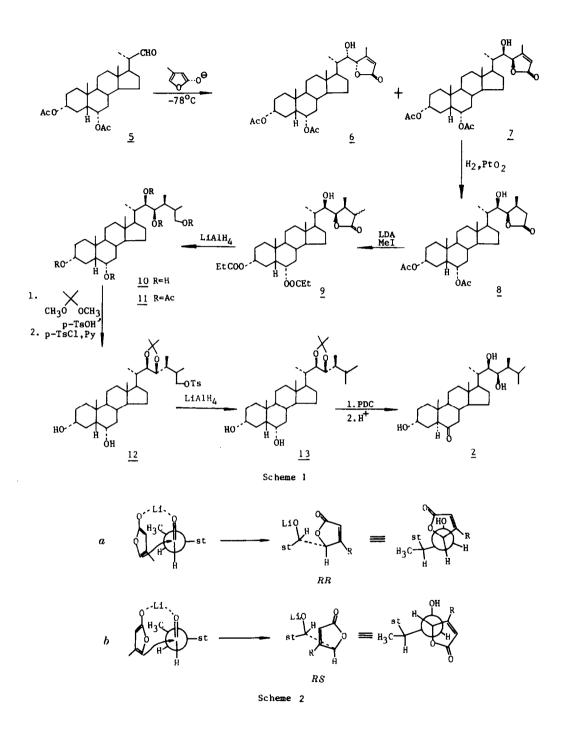
STUDY ON THE SYNTHESIS OF BRASSINOLIDE AND RELATED COMPOUNDS III STBREOSELECTIVE SYNTHESIS OF TYPHASTEROL FROM HYODEOXYCHOLIC ACID

WEI-SHAN ZHOU* and WEI-SHENG TIAN Shanghai Institute of Organic Chemistry, Academia Sinica 345 Lingling Lu, Shanghai, China (*Received in Japan* 19 *March* 1987)

Abstract----Typhasterol 2 was synthesized stereoselectively from hyodeoxycholic acid 3. The aldehyde 5 derived from 3 was reacted with the anion of 3-methylbutenolide under kinetic condition to afford (22R,23R)-7 as the major product. Hydrogenation of 7 and its 22-acetate 14 over PtO_2-Pt/C yielded the expected compound 8 and 15, respectively, in very good yield. Typhasterol was obtained from 8 and 15 through the following sequence of reaction $8 \rightarrow 9 \rightarrow 10$ $\rightarrow 12 \rightarrow 13 \rightarrow 2$ and $15 \rightarrow 16 \rightarrow 17 \rightarrow 13 \rightarrow 2$.

Since the discovery of novel plant-growth hormone brassinolide 1 in 1979^{1} , a number of new brassinosteroids in other higher plant have been isolated and identified². A simple analogue of brassinolide isolated from cat-tail pollen (Typha latifolia L.) represents the first example of 2-deoxy-brassinosteroid plant-growth hormone³. Although synthesis of brassinolide and its analogues has been achieved by several research groups from stigmasterol or ergosterol as starting material⁴, the synthesis of them starting from hyodeoxycholic acid 3 has not yet been reported. In our previous work, we have finished the conversion of A,B ring of hyodeoxycholic acid into the A,B ring of brassinolide⁵ and the synthesis of a stereoisomer of typhasterol 2, (225,235)-4 from 3^{6} . We now wish to report here the synthesis of natural typhasterol 2 also starting from 3.

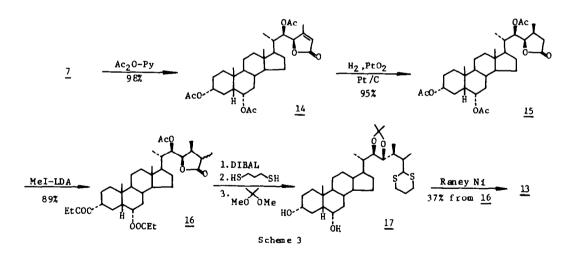




The aldol reaction of the aldehyde 5 derived from 3 with the anion of 3methylbutenolide generated in THF from butenolide and lithium diisopropylamide under kinetic condition yielded a mixture of the Cram 7 and anti-Cram 6 isomers in a ratio of 64:36 (Scheme 1)⁷. The coupling constant for H-20 to H-22 in major Cram isomer $7(J_{H(20)(22)}=0Hz)$ was much smaller than that of H-20 to H-22 in the minor anti-Cram isomer $6(J_{H(20)(22)}=4.4Hz)$ and the configuration at C-22 in 7 was therefore assigned as R^8 . The stereochemistry at C-23 seems to be determined by the approach of the anion as shown in scheme 2^{4d} . It is obvious that path a involves less steric interaction and is thus favored over path b, which results in the predominance of the 23 R stereochemistry. The configuration of C-23 was further confirmed by the CD curves(7: λ max 217nm, $\Delta \varepsilon = 11.9$; $6:\lambda$ max 227nm, $\Delta \varepsilon = -3.7$), which is in accord with that reported in literature⁹. Hydrogenation of 7 over PtO₂ yield 8 in 62% yield. The coupling constant for H-23 to H-24 in ¹H-NMR spectra of 8 is larger than 6Hz, and appears to be cis relationship for H-23 to H-24. Thus the configuration of C-24 was assigned as S¹⁰.

Methylation of **8** with CH_3I in the presence of LDA occurred not only at C-25 but also at the A-carbon of two acetoxyl groups of C-3 and C-6 to give **9** in 898 yield. Reduction of **9** with LiAlH₄ in THF afforded pentahydroxy compound **10**. Acetonation of **10** with 2,2-dimethoxypropane followed by selective tosylation with p-toluenesulfonyl chloride and reduction with LiAlH₄ afforded **13**. Conversion of **13** to typhasterol **2** was achieved in 408 yield utilizing procedure similar to those reported by us^{5,6}. The spectral data of **2** were identical with those reportd in litereture³.

An improved route for the synthesis of 13 was achieved from 7 through the following sequence of reaction: $7 \rightarrow 14 \rightarrow 15 \rightarrow 16 \rightarrow 17 \rightarrow 13$ in 30% overall yield from 7 which is much better than that obtained from the above route: $7 \rightarrow 8 \rightarrow 9 \rightarrow 10 \rightarrow 12 \rightarrow 13$ in 7% overall yield from 7.



EXPERIMENTAL

The silica gel used for the chromatography was 100-200 or 200-300 mesh in size and GP_{254} , silica gel H, respectively. Iodine vapour and vanillin were used for colour developing. The m.p.s were uncorrected. The optical rotation was measured on Autopol III polarimeter. IR spectra were recorded as KCl disks on Zeiss-75 model spectrometer. ¹H-NMR spectra were recorded on Varian XL-200 (200MHz) spectrometers using TMS as an internal standard. The unit of δ was ppm. Mass spectra were run on JMS-01U and MAT 711 instruments. Elemental analyses were performed by Analytical Department of this Institute.

Condensation of 3x,6x-diacetoxyl 22,23-bisnor-cholanyl aldehyde 5 with 3-methylbut-2-enolide 5-+6+7:

To a solution of 3.5ml of diisopropylamine and trace of α, α' -bipyridine in 30ml of dry THF was added 5ml of n-BuLi (1.8M/hexane) at $-30\,^{\circ}\text{C}$ under N₂ with stirring. A deep red solution was formed immediately. The reaction temperature was kept at $-30\,^{\circ}\text{C}$ for 30min and then allowed to warm to R.T. for further 30min. This LDA solution was cooled to $-78\,^{\circ}\text{C}$, and a solution of 3-methylbut-2-enolide (1.6g) in 20ml of dry THF was added dropwise. After the red color was disappeared, a solution of aldehyde 5 (lg) in dry THF (50ml) was then added dropwise and the mixture was stirred for 7 h, and guenched with 10% HCl at $-78\,^{\circ}\text{C}$. After extraction with CH₂Cl₂, the combined organic extracts were washed with brine, dried over Na₂SO₄. Removal of the solvent in vacuo followed by chromatography on silica gel(eluted solvent: benzene:acetone=9:1 or petroleum ether:ethyl acetate =4:1) gave 680mg of 7 in 55% yield and 280mg of 6 in 31% yield.

 $(22R,23R)-7: \text{ m.p. } 194-195^{\circ}C; [\alpha]_{p}^{25}+47.2^{\circ}(c \ 0.14, \ CHCl_{3}); \ CD: \ \lambda_{max} (CH_{3}OH)$ $217nm(\Delta \varepsilon=11.87); \ IR(KCl): \ 3400(OH), \ 1760, \ 1640(a,\beta-unsaturated lactone), \ 1730, \ 1240(acetate) \ cm^{-1}; \ ^{1}H-NMR(200MHz \ CDCl_{3}): \ 0.44(1H, \ s, \ OH), \ 0.69(3H, \ s, \ 18-H), \ 0.98(3H, \ s, \ 19-H), \ 1.11(3H, \ d, \ J=6.6Hz, \ 21-H), \ 2.02, \ 2.05(6H, \ 2s, 2xCH_{3}CO), \ 2.11 \ (3H, \ s, \ 28-H), \ 3.86(1H, \ s, \ 22-H), \ 4.71(1H, \ m, \ 3-H), \ 4.77(1H, \ s, \ 23-H), \ 5.15(1H, \ m, \ 6-H), \ 5.88(1H, \ s, \ 25-H); \ MS \ m/z \ 531(M^++1), \ 471(M^+-CH_{3}CO_{2}), \ 410(M^+-2xCH_{3}CO_{2}H), \ 312(M^+-98-2xCH_{3}CO_{2}H), \ 98(\ 4.77); \ Found: \ C, \ 70.37, \ H, \ 8.80; \ Calcd. \ for \ C_{31}H_{46}O_{7}: \ C, \ 70.19, \ H, \ 8.75.$

 $(22S,23S)-6: m.p. 219-221^{\circ}C; [a]_{\mathfrak{P}}^{28}+55.1^{\circ}(c\ 0.96,\ CH_{3}Cl);\ CD:\ \chi max(CH_{3}OH)$ 227nm($\Delta\varepsilon$ =-3.68); IR(KCl): 3400(OH), 1760,1640(a, \beta-unsaturated lactone), 1730, 1240(acetate) cm⁻¹; ¹H-NMR(200MHz CDCl_{3}): 0.70(3H, s, 18-H), 0.99(3H, s, 19-H), 1.08(3H, d, J=7Hz, 21-H), 2.02, 2.05(6H, 2s, 2xCH_{3}CO), 2.10(3H, s, 28-H), 3.89 (1H, d, J=4.4Hz, 22-H), 4.71(1H, m, 3-H), 4.90(1H, s, 23-H), 5.16(1H, m, 6-H), 5.88(1H, m, 25-H); MS m/z: 531(M⁺+1), 470(M⁺-CH_{3}CO_{2}H), 410(M⁺-2xCH_{3}CO_{2}H), 312 (M⁺-98-2xCH_{3}CO_{2}H), 98(Q)^{+}; Found: C, 70.66, H, 8.88; Calcd. for C₃₁H₄₆O₇: C, 70.19, H, 8.75.

Hydrogenation of (22R, 23R)-7 with Adams' catalyst $7 \rightarrow 8$:

7(100mg) dissolved in glacial acetic acid (6ml) was hydrogenated over PtO₂ (5mg) at room temp. for 10h. The catalyst was filtered off and the filtrate was poured into cold water (10ml). The precipitate was purified by flash column chromatography on silica gel to give 8 (62mg) in 62% yield. m.p. 120-124°C $[\alpha]_{D}^{17}$ -12.2°(c 0.49, CHCl₃); IR(KCl): 3450(OH), 1764(lactone),1720,1240(acetate) cm⁻¹; ¹H-NMR(200MHz CDCl₃): 0.67(3H, s, 18-H), 0.91(3H, d, J=6Hz, 21-H), 0.96 (3H, s, 19-H), 1.09(3H, d, J=7Hz, 28-H), 2.01, 2.04(6H, 2s, 2xCH₃CO), 3.97 (1H, d, J=6Hz, 22-H), 4.41(1H, dd, J=6Hz, 6Hz, 23-H), 4.70(1H, m, 3-H), 5.41(1H, m, 6-H); MS m/z: 472(M⁺-CH₃CO₂H), 412(M⁺-2xCH₃CO₂H), 313(M⁺-CH₃CO₂H-CH₃CO₂-99), 99($\frac{1}{2}$).

Methylation of 8 with LDA and methyl iodide $8 \rightarrow 9$:

To a solution of diisopropylamine (1.2ml) and trace of 2,2'-dipyridine in 5ml of dry THF was added dropwise n-BuLi (2.4ml, 1.6M/ether) at -30°C under N2. It was stirred at $-30\,^{\circ}\text{C}$ for 30 min and at room temperature for further 30 min. A deep red solution of LDA was formed. It was cooled to $-78\,^{\circ}\mathrm{C}$ and a solution of 8(260mg) in THF(16ml) was added. The mixture was stirred for 30min. When the red color of solution was discharged, 0.6ml of methyl iodide was added slowly. The reaction was kept at $-78\,^{\circ}$ C for 2h and quenched by addition of 10% HCl. The mixture was extracted with CH₂Cl₂. The extract was washed with brine, dried over Na2SO4 and concentrated under reduced pressure. The residue was chromatographed on silica gel (eluted solvent: petroleum ether: ethyl acetate= 5:1) to give 9(250mg) in 83% yield. m.p. 236-237°C; [A]²⁵-18.4°(c 0.64, CHCl₃); IR(KCl): 3400(OH), 1720(lactone) cm⁻¹; ¹H-NMR(200MHz CDCl₃): 0.68(3H, s, 18-H), 0.99(3H, s, 19-H), 1.13(3H, d, J=4.6Hz, 21-H), 1.15(3H, d, J=4.8Hz, 28-H), 1.25(3H, d, J=7.2Hz, 26-H), 3.88(1H, d, J=4.4Hz, 22-H), 4.41(1H, dd, J=4.4Hz, 6.6Hz, 23-H), 4.70(1H, m, 3-H), 5.18(1H, m, 6-H); MS m/z: 576(M⁺), 501(M⁺-C₃H₆O₂), 313(M⁺-114 -2xC3H6O2), 114((,), 57(CH3CH2CO); Found: C, 70.90, H, 9.54; Calcd. for C34H5607 C, 70.83, H, 9.72.

Reduction of 9 with lithium aluminum hydride $9 \rightarrow 10 \rightarrow 11$:

LiAlH₄(120mg) was added to the solution of **9** (120mg) in dry THF (10ml). The mixture was refluxed for 3.5 h. After cooling to room temperature the excess of LiAlH₄ was destroyed by addition of ethyl acetate, and then the mixture was extracted with CH_2Cl_2 . The extract was worked up as usual to give 98mg of **10**. MS m/z: 476(M⁺+1), 449(M⁺-H₂O), 431(M⁺-2xH₂O).

30mg of 10 in pyridine(1ml) was treated with acetic anhydride(1ml) at room temp. to afford 11(33mg) in 75.8% yield. m.p. 155-157°C; ¹H-NMR(200MHz CD₃COCD₃): 0.74(3H, s, 18-H), 0.94(3H, d, J=6.8Hz, 21-H), 0.98(3H, d, J=7Hz, 28-H), 0.99(3H, s, 19-H), 1.04(3H, d, J=6.6Hz, 26-H), 1.97, 1.98, 1.99, 2.01 (15H, 4s, 5xCH₃CO), 4.01(2H, d, J=6Hz, 27-H), 4.32(1H, m, 3-H), 5.09(1H, m,6-H), 5.18(1H, d, J=9Hz, 22-H), 5.28(1H, d, J=9Hz, 23-H); MS m/z: 617(M⁺-CH₃CO₂); 557 (M⁺-CH₃CO₂-CH₃CO₂H),497(M⁺-2xCH₃CO₂H-CH₃CO₂). Conversion of 10 into 13:

To a solution of 10(100 mg) in $\text{CH}_2\text{Cl}_2(\text{lml})$ was added 2,2'-dimethoxylpropane (lml) and p-TsOH(10mg). The solution was stirred at room temp. for 2h and then concentrated in vacuo to remove the excess of 2,2'- dimethoxylpropane. To the crude acetonide dissolved in pyridine(lml) was added p-toluenesulfonyl chloride (50mg) at 0°C, standing lh. After stirring at room temp. for 3h the solution was poured onto cooled water, and the precipitate was reduced with LiAlH₄ in ether(6ml) at room temp. for 10h. Work up as usual, the crude product was purified by chromatography on silica gel to afford 13(15mg) in 14% overall yield from 10. m.p. 166-168°C; ¹H-NMR(200MHz CDCl₃): 0.64(3H, s, 18-H), 0.86(3H, d, J=7Hz, 26-H), 0.87(3H, d, J=8.2Hz, 28-H), 0.91(3H, s, 19-H), 0.94(3H, d, J=6.8Hz, 27-H), 0.97(3H, d, J=5.4Hz, 21-H), 1.34, 1.37(6H, 2s, (CH₃)₂C); MS m/z: 489 (M⁺-1), 475(M⁺-CH₃), 419(M⁺-171), 171(⁶/₄-).

Acetylation of 7 with acetic anhydride and pyridine $7 \rightarrow 14$:

Hydrogenation of 14 over $Pt/C-PtO_2$ 14-+15:

To a suspension of $PtO_2(10mg)$ and 10%Pt/C in 2ml of glacial acetic acid was added a solution of dioxane (4ml) containing 8 (126mg) at room temp. under H₂. The mixture was continually stirred for 20h, and the catalyst was filtered off and the solvent was concentrated under reduced pressure. The residue was recrystallized from diisopropyl ether to give 15(120mg) in 95% yield. m.p.211-213°C; $[\alpha]_{p}^{21}$ -11.87°(c 0.23, CH₃OH); IR(KC1): 1790(1actone), 1740, 1240(acetate) cm⁻¹; ¹H-NMR(200MHz CDC1₃): 0.68(3H, s, 18-H), 0.97(3H, s, 19-H), 1.03(3H, d, J=6.8Hz, 21-H), 1.11(3H, d, J=6.8Hz, 28-H), 4.50(1H, dd, J=8.4Hz, 6Hz, 23-H), 5.24(1H, d, J=8.4Hz 22-H); MS m/z: 575(M⁺+1), 515(M⁺-CH₃CO₂), 454(M⁺-2xCH₃CO₂H), 395(M⁺-2xCH₃CO₂H-CH₃CO₂-99), 99().

Methylation of 15 15→16:

Methylation of 15 was the same as that of 8 in yield of 89%. m.p. 94-98°C; ¹H-NMR(200MHz CDCl₃): 0.66(3H, s, 18-H), 0.97(1H, s, 19-H); MS m/z: 557(M⁺-CH₃-CH₂CO₂), 482(M⁺-CH₃CH₂CO₂H-CH₃CH₂CO₂), 409(M⁺-CH₃CH₂CO₂H-2xCH₃CH₂CO₂).

Conversion of 16 into 17:

To a solution of 16 (74mg) in toluene (2ml) was added dropwise 0.4ml of the diisobutyl aluminum hydride (DIBAL, 1M/toluene) at- 78°C under N₂. The solution was kept at -78°C for 5h. The solution was allowed to come to room temp., and poured onto the saturated NH₄Cl and extracted with CH₂Cl₂. The extract was worked as usual manner to give 64mg of product. IR(KCl): 3400(OH) cm⁻¹.

The product dissolved in 4ml of CH_2Cl_2 was treated with $TiCl_4$ (0.4ml) and propane 1,3-dithiol. The mixture was stirred at room temp. for 27h and extracted with CH_2Cl_2 . The extract was washed with 5% NaOH, brine and dried over Na_2SO_4 . After removal of the solvent, the residue was treated with 4ml of 2,2'-dimethoxy propane in the presence of p-TsOH(10mg) at room temp. for 8h. Work up gave 80mg of 17 which could be used directly for next step without further purification. ¹H-NMR(200MHz CDCl_3): 0.64(3H, s, 18-H), 0.91(3H, s, 19-H), 0.93(3H, d, J=7.4Hz, 21-H), 0.97(3H, d, J=7.4Hz, 28-H), 1.09(3H, d, J=6.8Hz, 26-H), 1.35, 1.38(6H,2s, (CH₃)₂C), 2.88(4H, m, -SCH₂CH₂CH₂S-), 4.11(1H, d, J=6Hz, -SCHS-); MS m/z: 594 (M⁺), 579(M⁺-CH₃), 173(\checkmark_{5}^{5})¹, 146(\searrow_{5}^{5})¹, 119(+ \langle_{5}^{5}).

Hydrogenolysis of 17 over Raney Ni 17-+13:

To a suspension of Raney Ni (10mg) in 10ml of absolute ethanol was added a solution of 17 (80mg) in ethanol under H_2 . The mixture was refluxed for 9h and the catalyst was filtered and the solvent was removed under reduced pressure. The residue was chromatographed on silica gel to give 13 (21mg) in 37% overall yield from 16. m.p. 166-168°C. The m.p. of this compound was not depressed on mixing with that obtained in the previous experiment. ¹H-NMR(200MHz CDCl₃): 0.64(3H. s, 18-H), 0.86(3H,d, J=7Hz, 26-H), 0.87(3H, d, J=8.2Hz, 28-H), 0.91 (3H, s, 19-H), 0.94(3H, d, J=6.8Hz, 27-H), 0.97(3H, d, J=5.4Hz, 21-H), 1.34, 1.37(6H, 2s, (CH₃)₂C); MS m/z: 489(M⁺-1), 475(M⁺-CH₃), 419(M⁺-171), 171($\frac{1}{2}$,).

Typhasterol 2:

To a solution of 13(18mg) in CH₂Cl₂(2.5ml) was added PDC(20mg). After the solution was stirred at room temp. for 2.5h, acetone(5ml) was added and filtered. The filtrate was concentrated under reduced pressure to give the residue which was dissolved in 5% HCl-MeOH (5ml) and allowed to stand overnight followed by chromatography on silica gel to afford 2 (3mg). m.p. 225-228°C(CH₃OH-H₂O)(Lit³ m.p. 227-230°C); ¹H-NMR(400MHz,CDCl₃): 0.68(3H, s, 18-H), 0.72(3H, s, 19-H), 0.74(3H, d, J=7Hz, 28-H), 0.80(3H, d, J=6.9Hz, 21-H), 0.93(3H, d, J=6.9Hz, 26-H), 1.03(3H, d, J=6.9Hz, 27-H), 1.74(2H, dd, J=8.3Hz, 2.5Hz, 4-H), 2.30(1H, dd,J=13Hz, 4.5Hz,7β-H), 2.61(1H, t, J=8Hz, 54-H), 3.44(1H, d, J=9.3Hz, 22-H), 3.68(1H, d, J=4.1Hz, 23-H), 4.15(1H, Wy=6.5Hz, 3-H); HRMS m/z: 449.3589(M⁺+1, requires 449.3631, C₂₈H₄₉O₄), 348.2609(M⁺ - $\frac{+1}{OM}$, requires 348.2665 C₂₂H₂₆O₃), 329.2460 (M⁺ - $\frac{+1}{OM}$ - H₂O, requires 329.2482, C₂₂H₃₃O₂), 271.2039(M⁺ - $\frac{+1}{OM}$ - H₂O, requires 271.2062, C₁₉H₂₇O).

References and Notes

- M.D.Grave, G.F.Spencer, W.K.Rowedder, N.B.Mandava, J.F.Worly, J.D.Jr.Warthen G.J.Steffens, J.L.Flippen-Anderson and J.C.Jr.Cook, Nature <u>281</u> 216(1979).
- a. T.Yokota, M.Arima, N.Takahashi, Tetrahedron Lett., <u>23</u> 1275(1982); b. T. Yokota, J.Baba, N. Takahashi, Tetrahedron Lett., <u>23</u> 4965(1982); C. T.Yokota, M.Morita, N.Takahashi, Agric. Biol. Chem., <u>47</u> 2149(1983); d, S.Takatsuto, N. Ikekawa, H.Abe, T.Morishita, M.Uchiyama, M.Ikeda,T.Sasa, S.Marumo, T.Kitsuwa, 25th Symposium on the Chemistry of Natural Product, Tokyo, 1982, p.290. e. H.Abe, T.Morishita, M.Uchiyama, S.Takatsuto, N.Ikekawa, Agric. Biol. Chem., <u>48</u> 2171(1984).
- a.J.A.Scheider, K.Yoshihara, K.Nakanishi, N.Kato, Tetrahedron Lett., <u>24</u> 3895 (1983); b. S.Takatsuto, N.Yazawa, M.Ishiguro, M.Morisaki and N.Ikekawa, J. Chem. Soc. Perkin I, 139(1984).
- 4. a. S.Fung, J.B.Siddall, J. Am. Chem. Soc., <u>102</u> 6580(1980); b. S.Takatsuto, N.Ikekawa, J. Chem. Soc. Perkin I, 2133(1983); c. K.Mori, M.Sakakibara, K. Okada, Tetrahedron <u>40</u> 1767(1984); d. J.R.Donaubauer, A.M.Greaves, T.C.McMorris, J. Org. Chem. <u>49</u> 2833(1984); e. M.Anastasia, P.Allevi, P.Ciuffreda, A.Fiecchi, A.Scalca, J. Org. Chem., <u>49</u> 4297(1984); f. T.Kametani, T.Katoh, M.Tsubuki, T. Honda, J. Am. Chem. Soc. <u>108</u> 7055(1986).
- 5. W.S.Zhou, W.S.Tian, Acta Chimica Sinica, 42 1173(1984).
- 6. W.S.Zhou, W.S.Tian, Acta Chimica Sinica, 43 1060(1985).
- 7. W.S.Zhou, W.S.Tian, H.Zheng, X.Y.Zhao, Acta Chimica Sinica, to be published. In this paper, the relationship between stereochemistry and temperature in the aldol reaction of the steroidal aldehyde 5 with the 3-methylbut-2-enolide anion has been studied. The most suitable reaction temperature(-78°C) has been found in this aldol reaction to give 7 and 6 in the ratio of 70:30 in 98.7% yield.
- 8. M.Sakakibara, K.Mori, Agric. Biol. Chem., 47 663(1983).
- 9.J.A.Edwards, J.Sundeen, W.Salmond, T.Iwadare, J.H.Fried, Tetrahedron Lett., 791(1972).
- 10.X.T.Liang "NMR, the resolution and application of higher resoluted ¹H-NMR spectrum" Science Press, Beijing 1976, p.296.

Acknowledgement:

This investigation was supported by the Science Funds of the Chinese Academy of Sciences.